

One-atom-thick Membrane for Advanced Separation

Vice-Chancellor's Research Day

20th January 2025

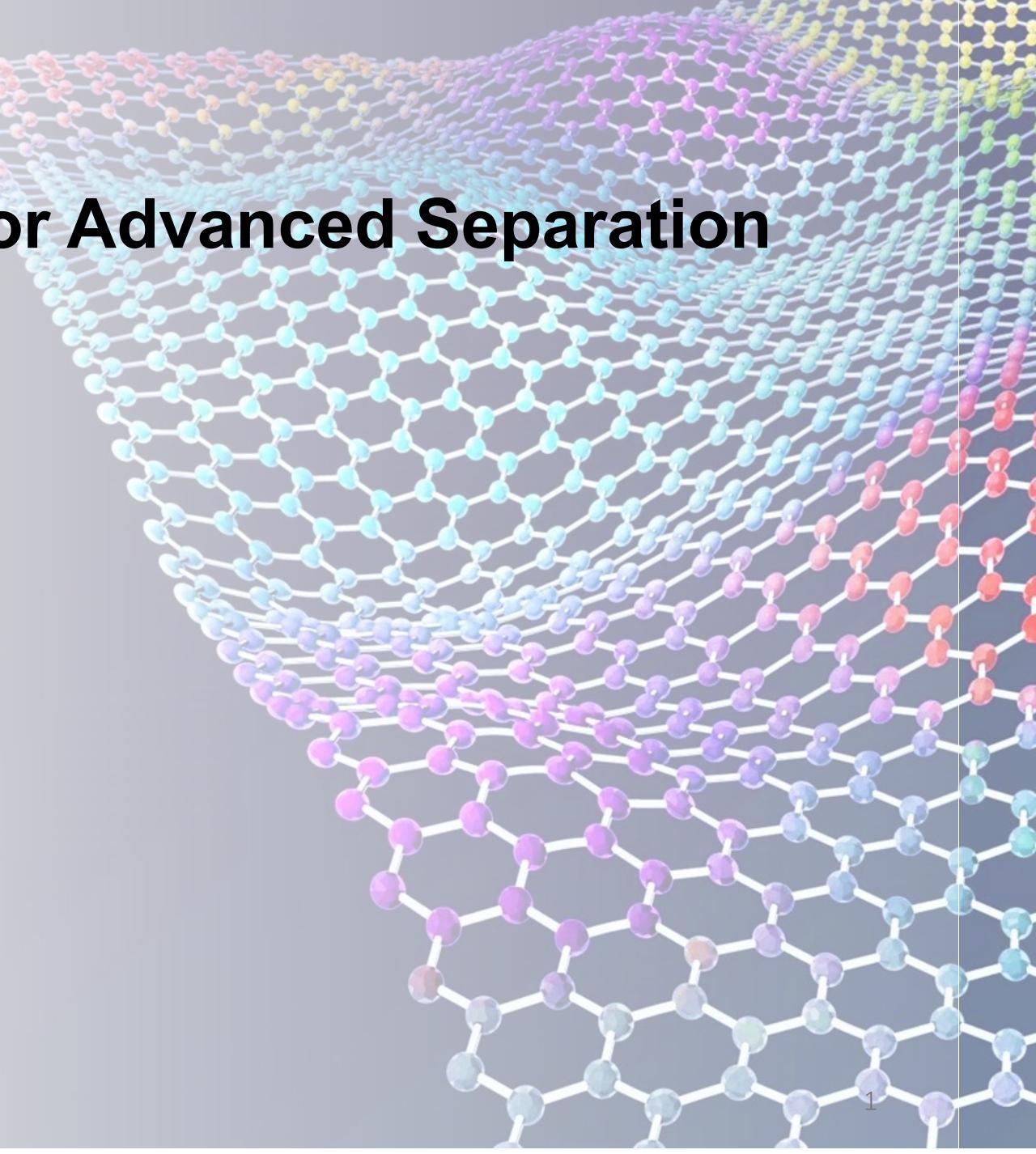
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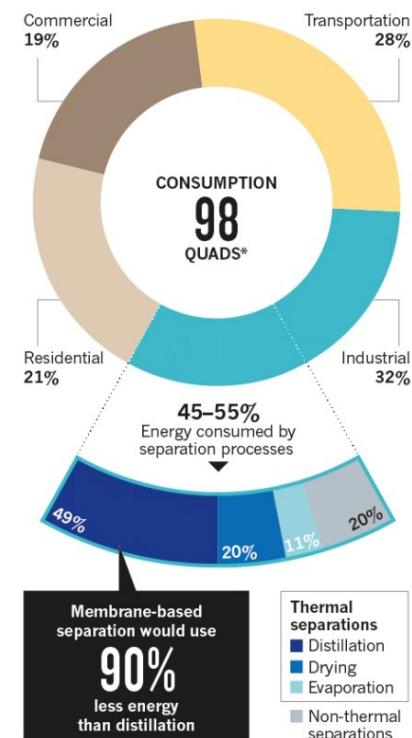
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Motivation

Industrial De-carbonization Target

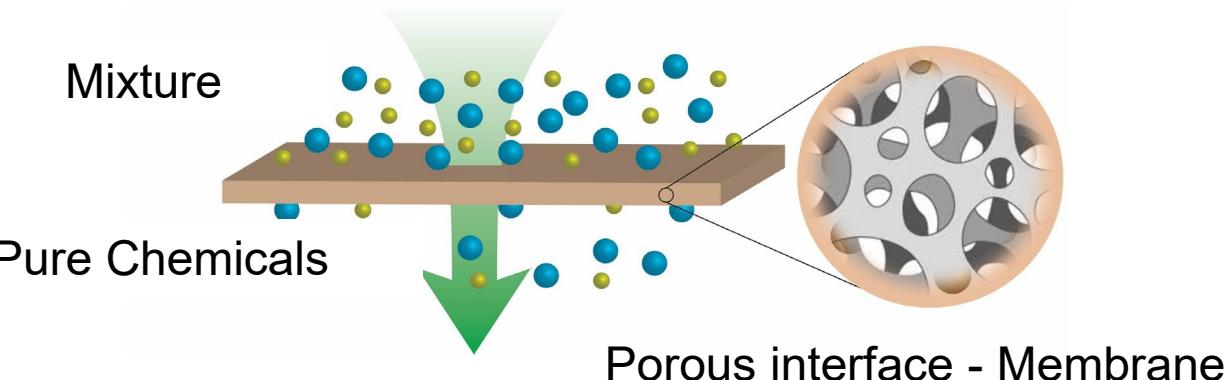


By 2030

- ☐ Switch to 20 TWh of low-carbon fuels (UK)
- ☐ Capture 3 million tons of CO₂ (UK)

Energy-Efficient Separation Process

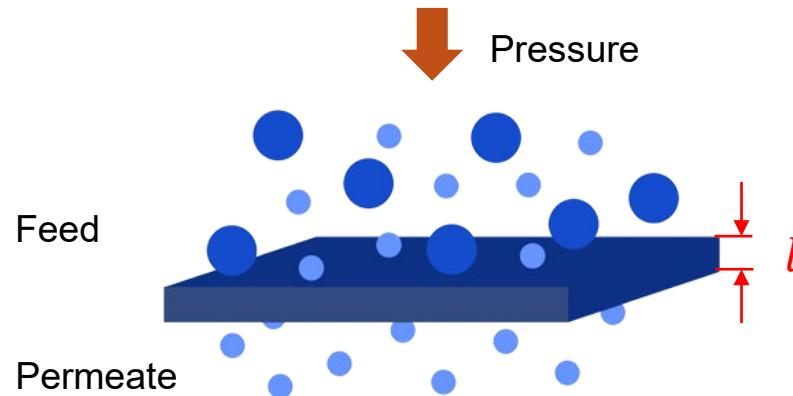
Separation Energy consumption:
6 million tonnes oil/year for UK



Membrane is promising technology to

- ☐ Save 90% separation energy
- ☐ Cut down Carbon capture cost < 20\$/ton

My Research focus – one-atom-thick membranes

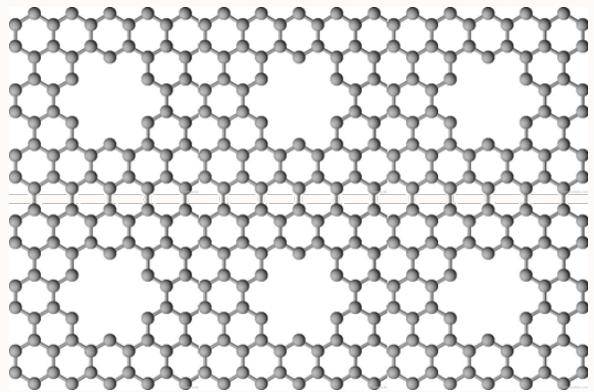


Properties

- ✓ One-atom-thick layer (e.g. graphene)
- ✓ High carrier (electron) mobility
- ✓ Unique optical property

Transform one-atom-thick films as high-performance membranes.

□ Shortest mass transport channel



High performance membrane

□ Unique electronic/optic property

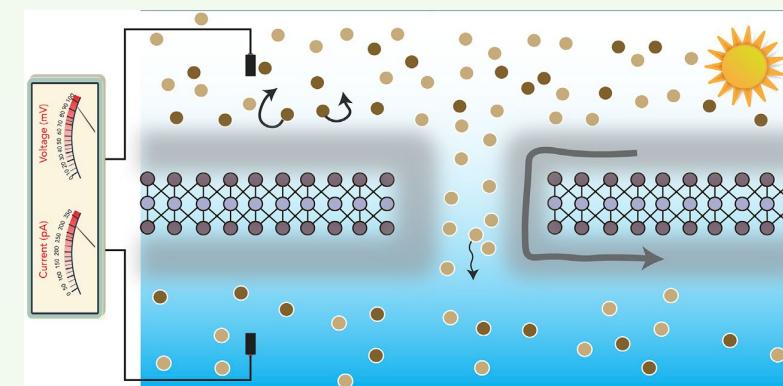
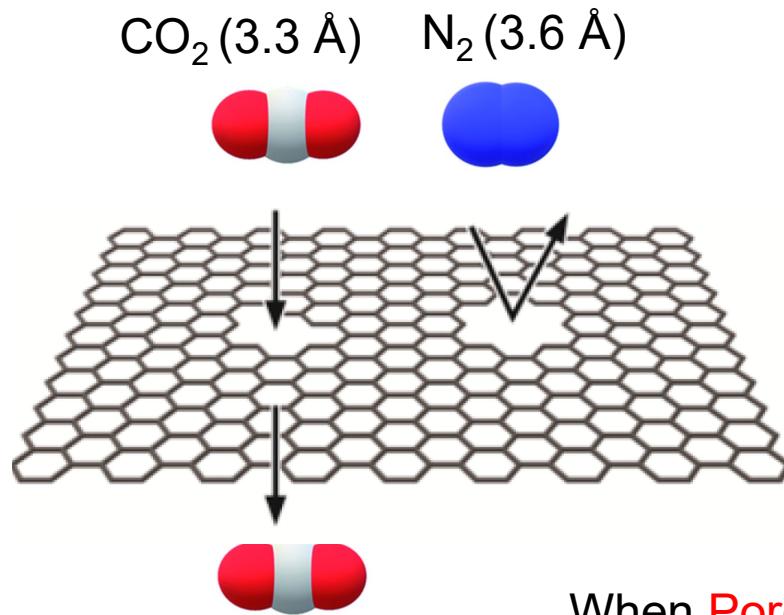


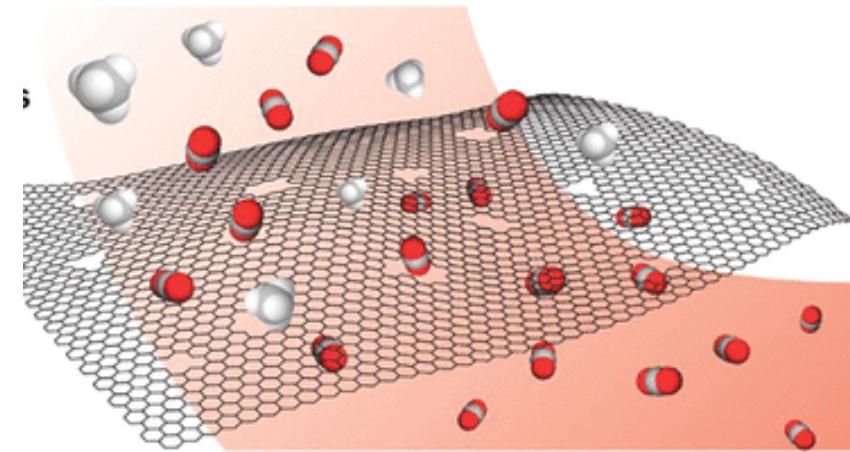
Photo-enhanced mass transport

Challenges to incorporate sub-nm nanopores

High resolution nanopore



High density nanopores



When **Pore density** reach 10^{12} cm^{-2} to 10^{13} cm^{-2}

$$1 \text{ \AA} = 10^{-10} \text{ m}$$

Expected 10^4 to 10^5 GPU gas permeance

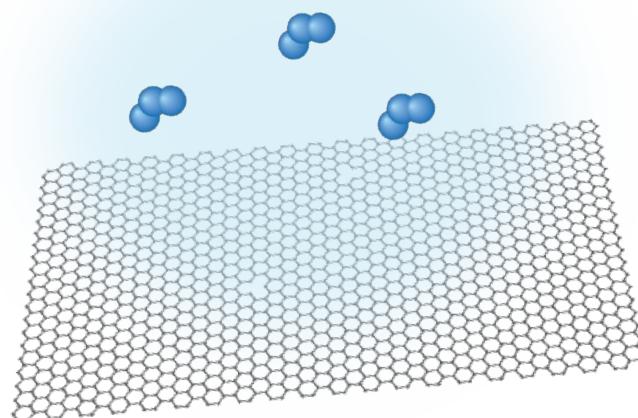
≈ 10 to 100 times of commercial membranes

Our approaches

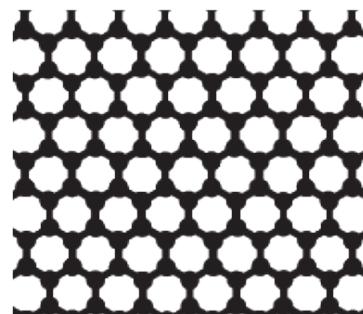
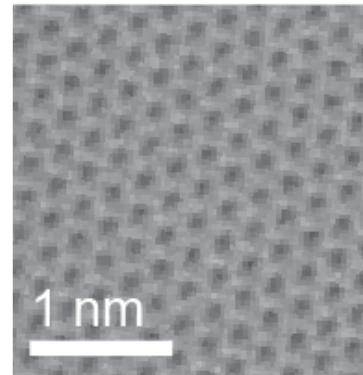
Nanopore
engineering

Mathematical
modeling

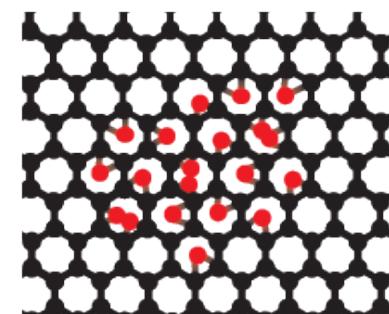
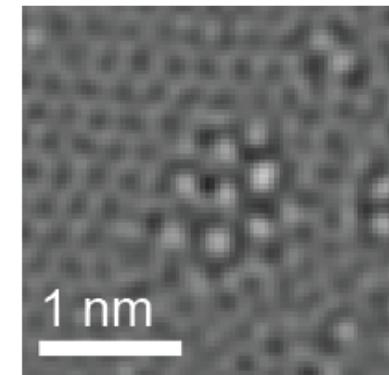
Energy-efficient
Membrane



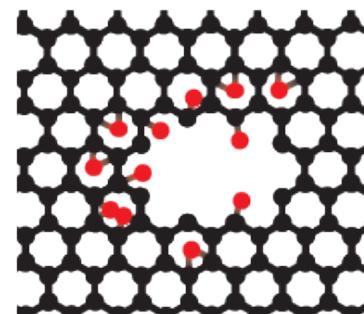
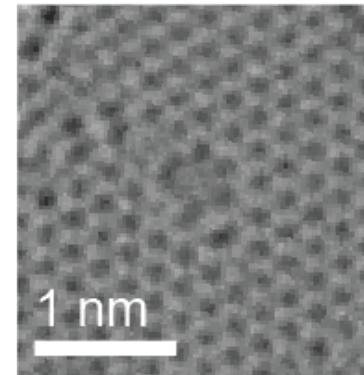
Ozone etching



Pristine Graphene

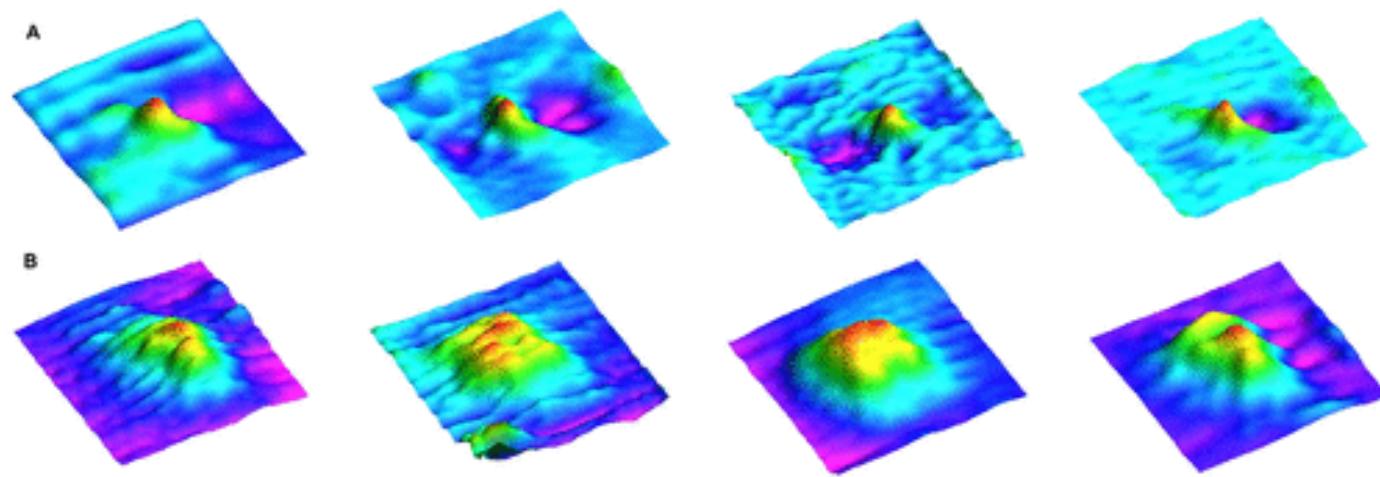


Functionalized
Graphene

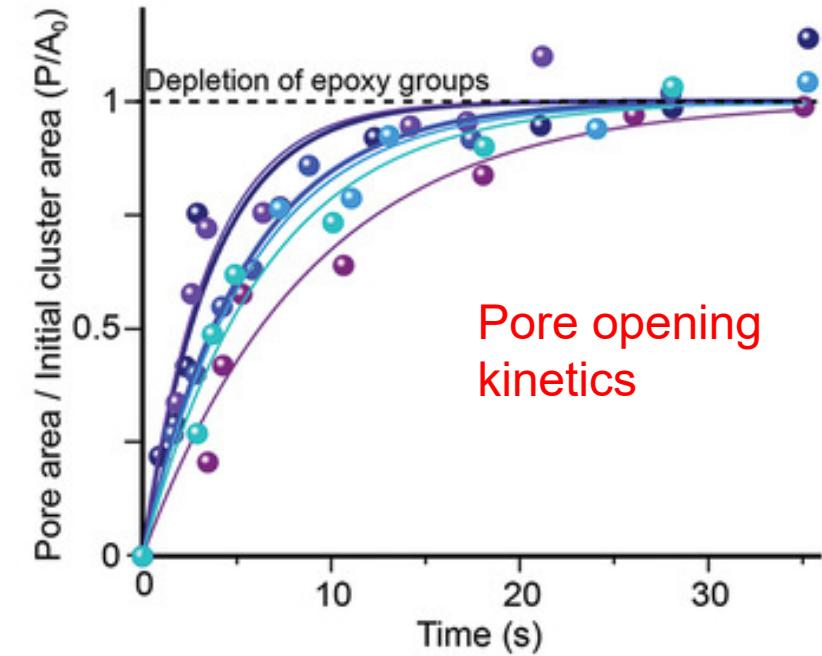


Nanopore
formation

1. In-situ probe nanopore etching kinetics

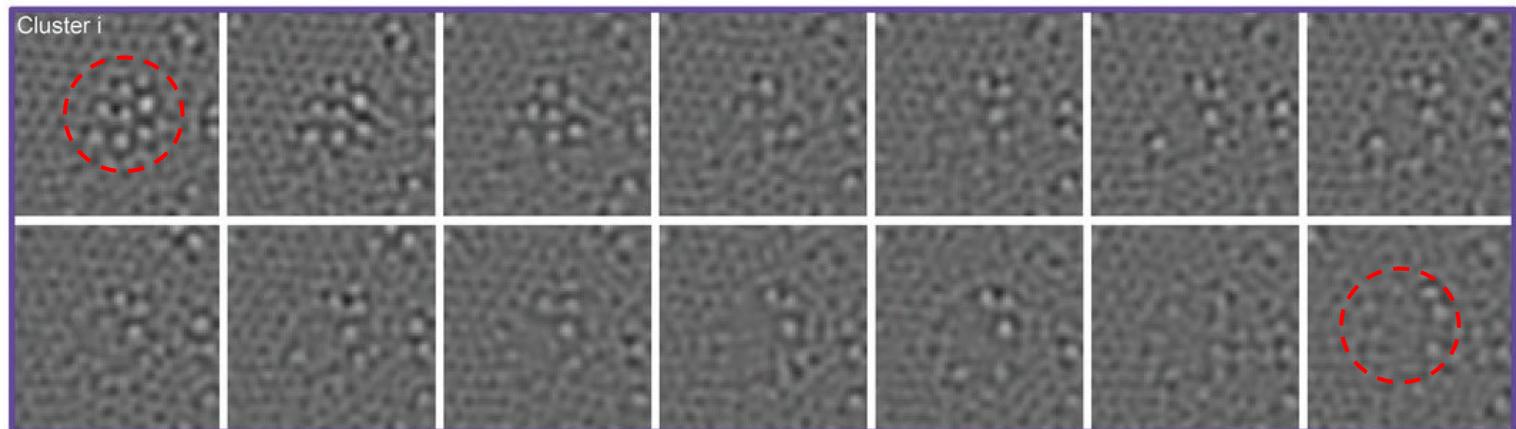


3D STM to illustrate the functional cluster formation



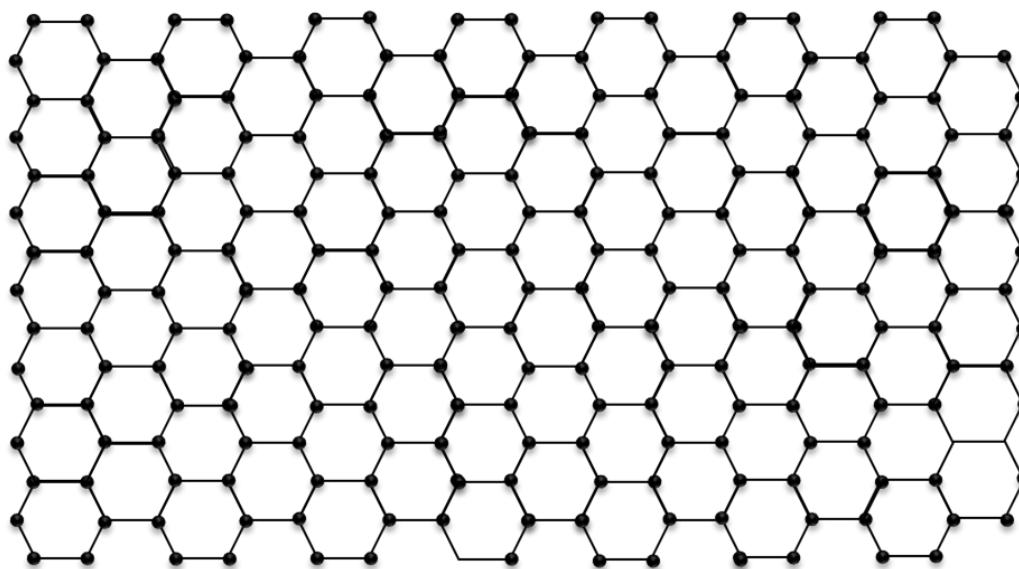
O-clusters in graphene lattice:
1 h exposure to 10 % O_3 at 43 °C

O-cluster to pore:
80-keV beam with a dose rate of
15700 $e^- s^{-1} \text{\AA}^{-2}$



AC-HRTEM to probe the nanopore opening

2. Mathematical modelling to guide nanopore formation

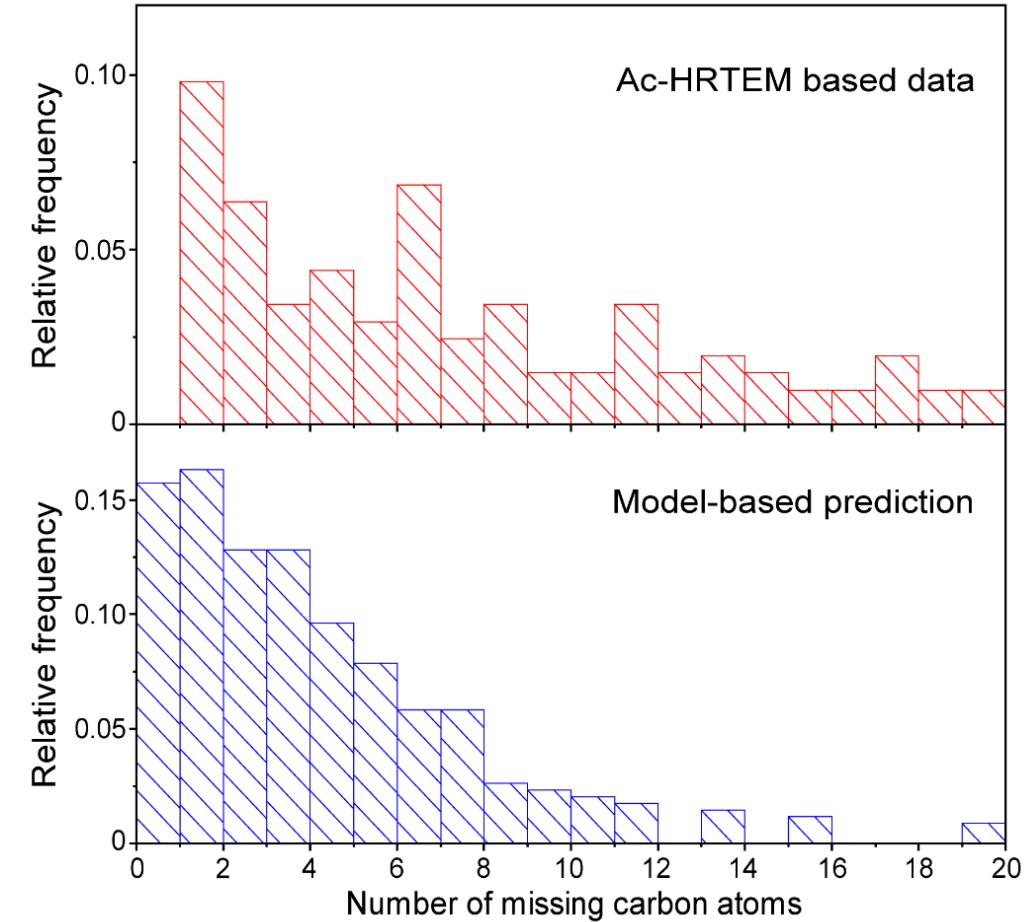


Nucleation sites during $[t_{i-1}, t_i]$:

$$N_i = C_{\text{density}} k_c H \int_{t_{i-1}}^{t_i} P dt$$

Carbon atom removing of the pores nucleated during $[t_{i-1}, t_i]$.

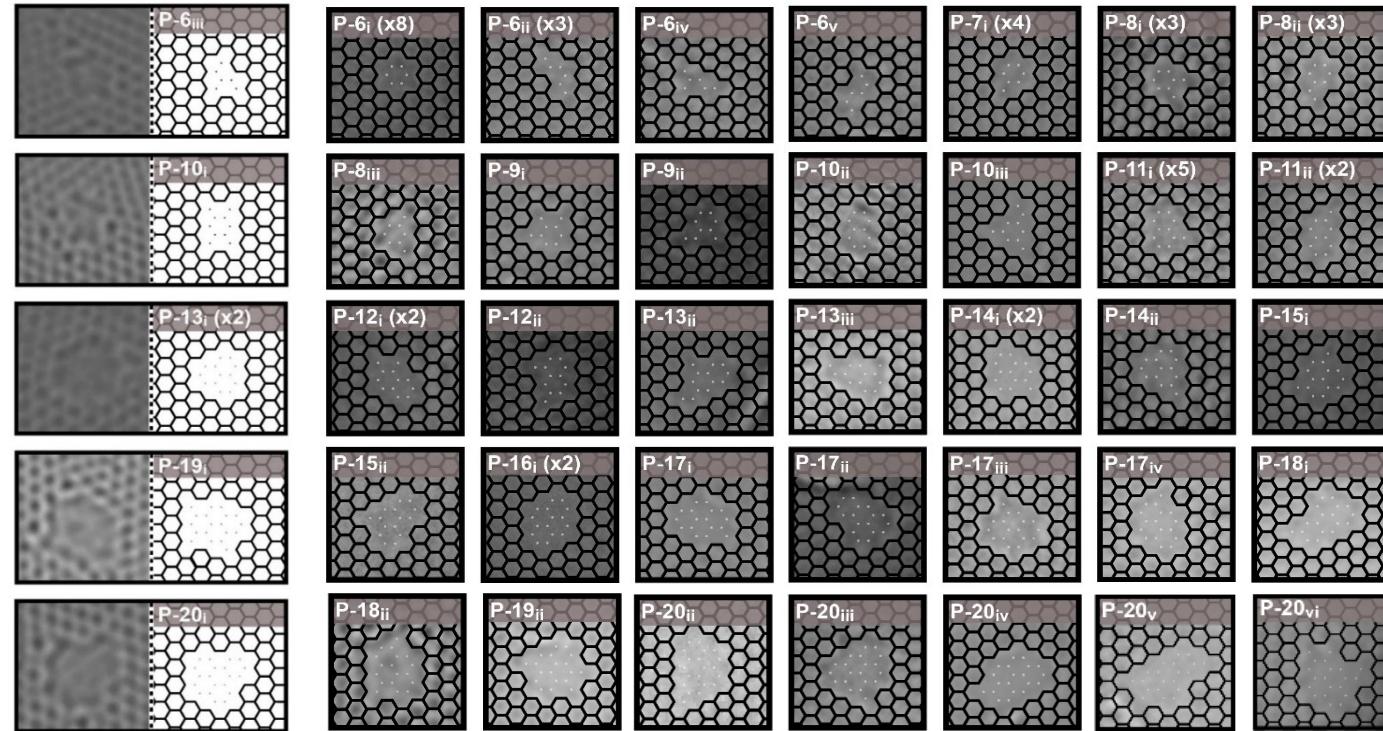
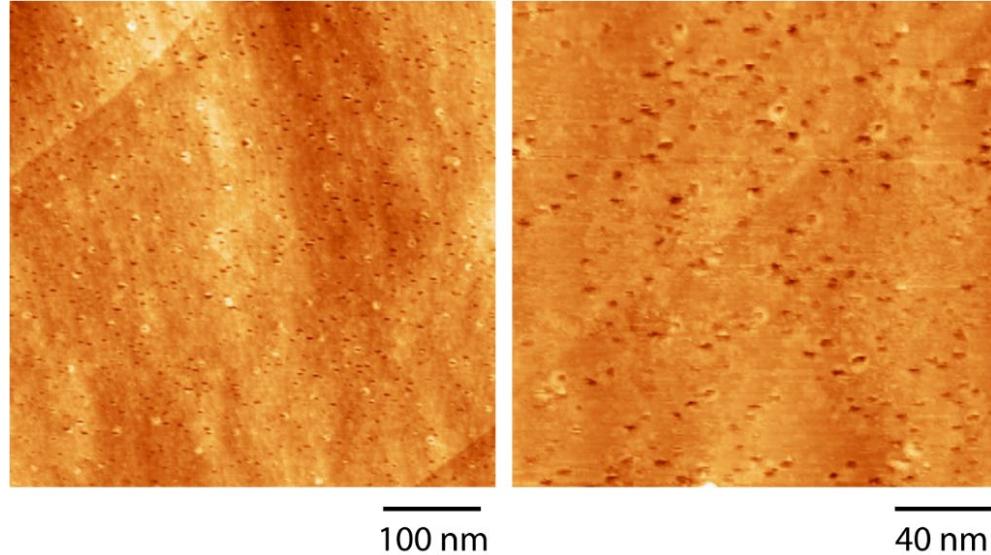
$$v_i = \sum_{m=i}^{m=n} \frac{\Delta C_m}{\sum_{k=1}^m N_k} = \frac{\Delta C_i}{N_1 + N_2 + \dots + N_i} + \dots + \frac{\Delta C_n}{N_1 + N_2 + \dots + N_n}$$



We design our nanopores!

2. Mathematical modelling to guide nanopore formation

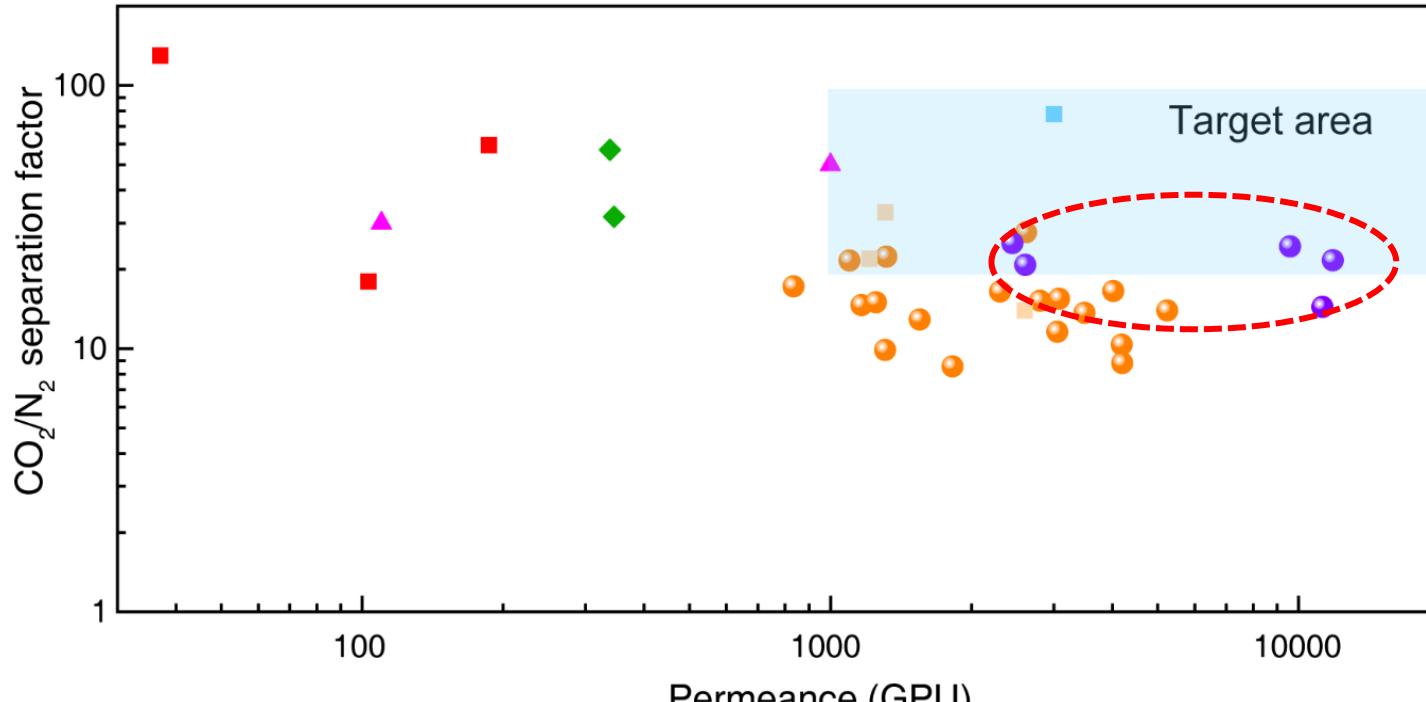
Based on mathematical modelling, optimized experiment condition was chosen to form the following nanopores



Realize uniform sub-nanometer pore (6 – 20 missing carbon atoms) with high density $1.6 \times 10^{12} \text{ cm}^{-2}$

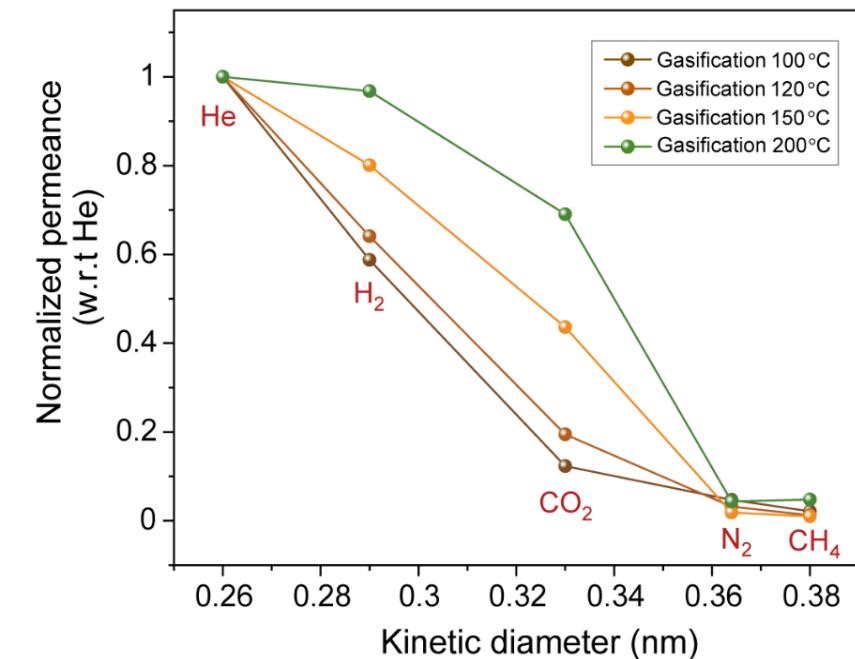
Some of structure observed in experimentally for the first time!

3. The 1st reported one-atom-thick membrane for gas separation

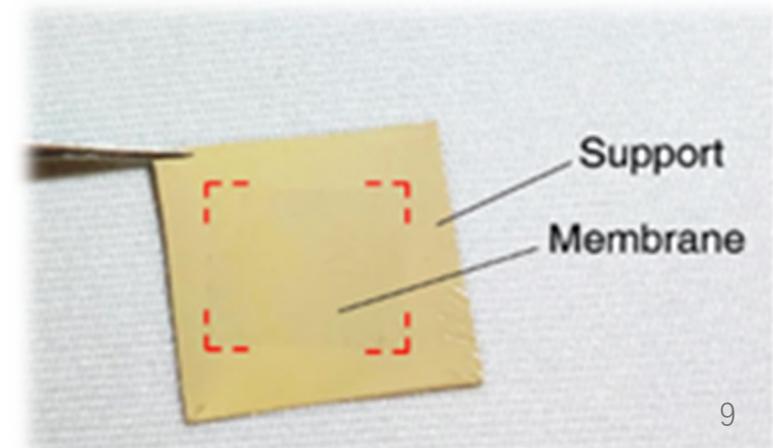


High permeance membrane for CO_2/N_2

Within 1% porosity
10 times higher permeance than commercial membrane



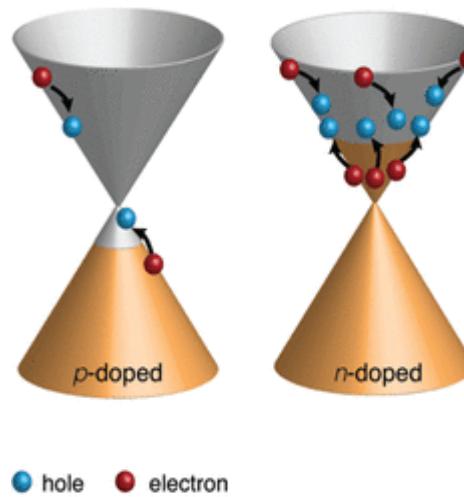
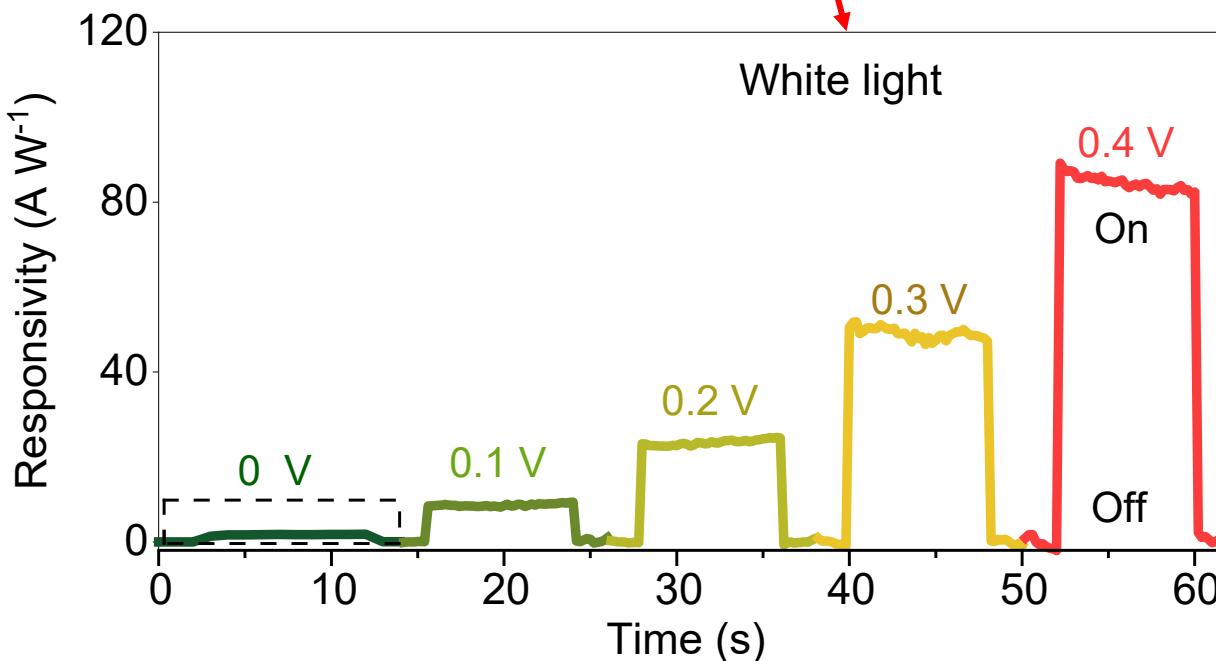
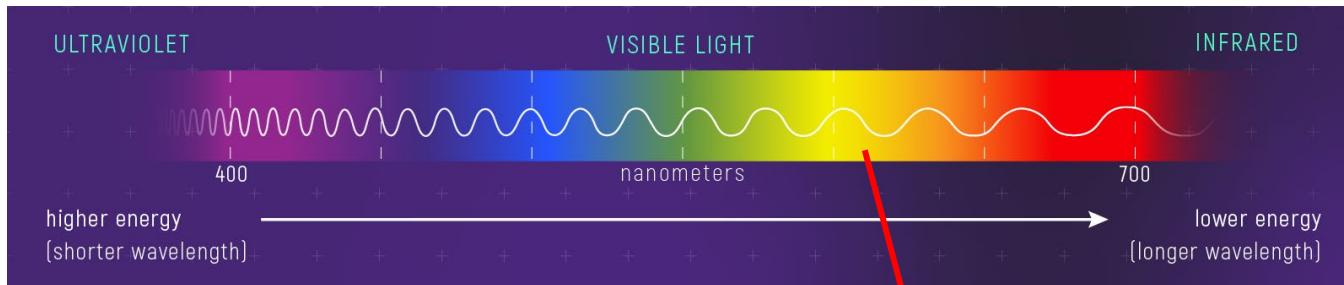
Fine tune for different gas pairs



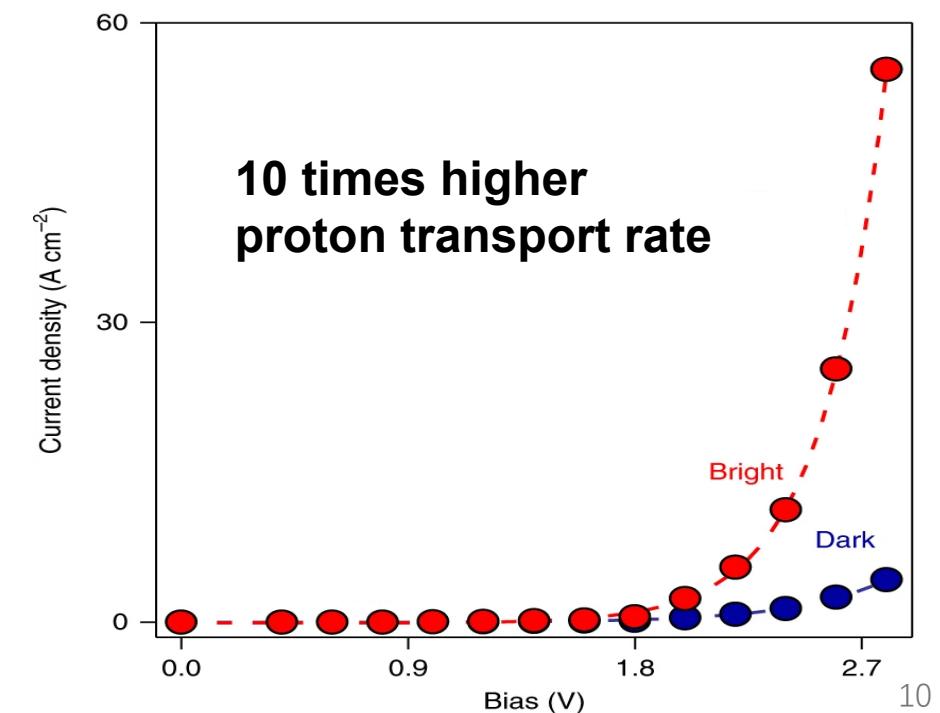
Can we save more energy?

Photo-enhanced graphene membrane

Mass transport ~ Electron structure ~ Photon effect



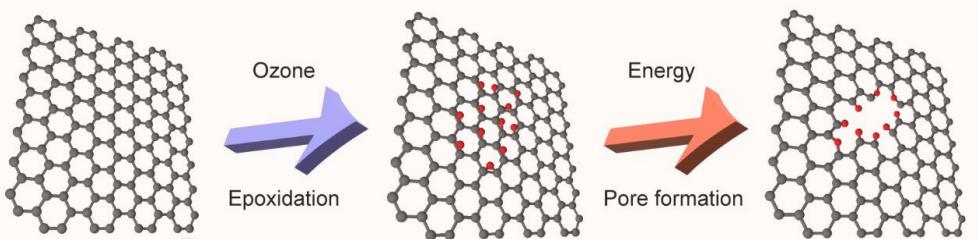
Photons ~ Electrons



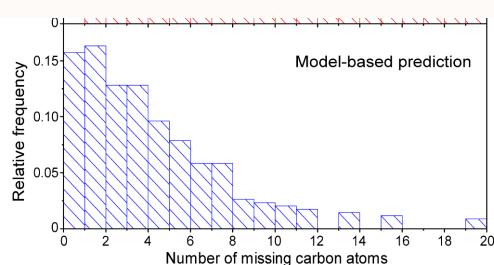
Summary

Engineering one-atom-thick graphene as a high-efficiency mass transport interface

Nano Engineering to high performance separation

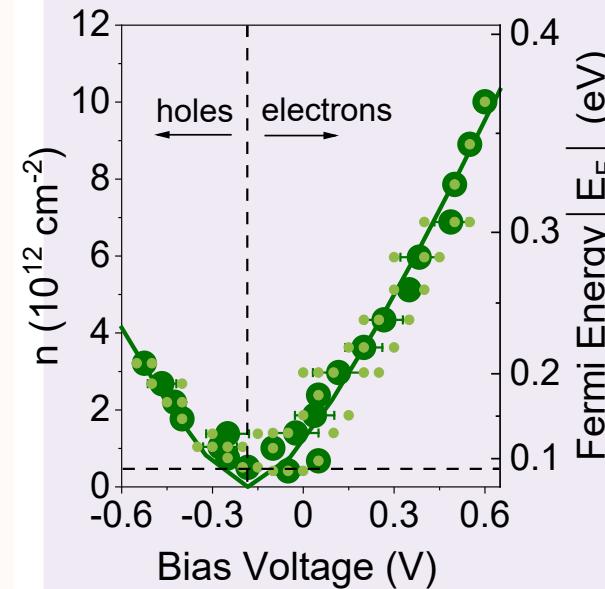


In-situ probe nanopore formation



Mathematical model-guided nanopore fabrication

Unique photo-enhanced phenomenon



Tune Fermi energy control photo-excited electrons

Acknowledgment



Thank you for your attention!

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